

Utilization of Manganese clay industrial by-product IN BUILDING MATERIAL INDUSTRY

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Introduction

Building materials (BM) in general

Role of building materials in human health, radiological aspect

Raw and add materials

New synthetic BM



Reuse of industrial by-products (BP)

• Potential starting material

Recycling could reduce the environmental impact







Overall radiological survey of NORM origin by-products

Favorable compounds \rightarrow NORM by-products as raw or add material in "clay-based" BM industry

- Red mud
- Coal slag
- Manganese clay
- Oil sludge

Mixing with clay matrix

Survey of potential Hungarian clays used in building material production

Unified classification protocol in PE-RRI



Reuse possibility of NORMs in BM production Decorative color as add material or raw material in case of brick production

Closest building material factory capacity: 200 million brick/year ightarrow 600 000 t/year

- Distance of Manganese clay reservoirs: 22 km
- 20% Manganese clay: 120 000 t/year
- Total Mn-clay = 2 800 000 t → 23 year
- Distance of **red mud** reservoirs: 7 km!
- 20 % red mud: 120 000 t/year
- Total Red mud = 50 000 000 t \rightarrow ~400 year







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Manganese clay

- Originated as a result of manganese mining and processing of oxide type Manganese ore
- The ore was separated from the clay with excelsior (washed)
- •~2.8 Mt of manganese clay have deposited
- Favorable matrix features
- Ion exchanger capacity





Manganese clay

Chemical components [%]								
SiO ₂	SiO₂ 29-33 BaO 0.05-0.							
TiO ₂	0.3-0.4	CaO	3-7					
Al ₂ O ₃	6-10	MgO	2-4					
Fe ₂ O ₃	22-26	K ₂ O	2-3					
MnO ₂	13-19	Na ₂ O	0.2-0.3					
MnO ₂	2-3	P_2O_5	0.4-0.5					



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Radionuclide content

Natural radioisotope content

Manganese (oxide, carbonate) ore, → Natural origin radioisotopes (U-238 (Ra-226): 20-200 Bq/kg, Th-232: 40-400 Bq/kg, 600-900 Bq/kg)

Radiological aspect of natural isotopes in case of BMs







Classification



Gamma-dose (Ra-226, Th-232, K-40)

• I-index

$$I = \frac{C_{Ra-226}}{300} + \frac{C_{Th-232}}{200} + \frac{C_{K-40}}{3000}$$

Material	Dose criterion (mSv y ⁻¹) <u>1.0</u>		
Used in bulk amounts	1<10		
(concrete, brick, etc.)	<u>1 3 1.0</u>		
Superficial or with			
restricted use (tile, etc.)	1 ≥ 0.0		

Radon exhalation

Problematic task

Radon exhalation greatly depends on the structure of the materials

The investigation of possibilities to reduce exhalation capacity are very important

• Radon radon level \rightarrow 300 Bq/m³

Leaching of radionuclides

Leaching tests are required





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Classification protocol in PE-RRI

Determination of radionuclide content

- By-product sampling
- Gamma spectrometry, alpha spectrometry
- Classification (I-index)

Radon emanation exhalation influencing effects

- Heat-treatment
- Moisture content
- Sample thickness

Internal structure features

- Porosity
- Superficial morphology

Leaching behavior tests

- CEN/TS 15364:2006 Characterization of waste
 - Acid and base neutralization capacity test





Measurements and methods

SAMPLE COLLECTION

Clay samples

Manganese clay

- Clay samples were taken from
 20 sample from Manganese Tüskevár (Hungary)
 - clay reservoir (Úrkút)
 - 0 40 cm depth

GAMMA SPECTROMETRY

- Drying, milling, storage in Marinelli vessels
- Instrument: ORTEC GMX40-76 HPGe detector with efficiency of 42 %
- Data collection: Tennelec PCA-MR 8196 MCA
- Measurement time: 80 000s







Determination of radon emanation and exhalation influencing parameters

Spherical and stick shape (d = 4 - 5 mm) clay-mixed samples (20% Manganese clay)

Effect of heat-treatment

- Preheated kiln
- 4 h heat-treatment 100 750 °C

Exhalation measurement

- Optimal measurement conditions (Thickness \rightarrow free exhalation state)
- Effect of heat-treatment
- Internal structure measurements
- Specific surface, porosity (focused on micro and mezo pores)
- Pore diameter





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Leaching tests



EN/TS 15364:2006 (ANC/BNC tests)

- For waste categorization
- Acid and base neutralizing capacity at 8 points between pH 4-12
- Liquid/solid ratio, concentrations, pH, redox potential, complex forming capacity and the aging of the waste can be measured

Provide information about the long term behavior of the waste

MSZ-21470-50

Hungarian regulation for toxic elements, heavy metals and chrome(IV) in soil

- 4 one-step extractions
- distilled water water soluble
- Lakanen-Erviö solution available for plants
- $HNO_3+H_2O_2$ total digestion
- HNO₃+HCl total digestion



Leaching tests



Tessier-extraction

For metals

- 5 step sequential extraction
- Gives information about the speciation

Ion-exchangable	1 M MgCl ₂ at pH 7 for 1 h at room temperature with constant stirring			
Bound to carbonates	NaOAc set to pH 5 with acetic acid at room temperature for 5 h with constant stirring			
Bound to Fe- and Mn-oxides	0.04 M NH ₂ OH*HCl in 25% acetic acid for 6 h at 93 ^o C			
Bound to organic matter	0.02 M HNO3 and 30% H2O2 set to pH2 with HNO3 for 4 h with occasional stirring, then 3,2 M NH4OAc in 20% HNO3, dilute with distilled water and 30 minute of vigorous stirring			
Residual	Microwave digestion			



Mesurements of leachate



Na, K, Ca Mg, Heavy metals, rare earth: -ICP, ICP-MS

NORM nuclides:

- Ra isotopes (226, 224, 228) alpha spectrometry + Rn emanation method
- U isotopes (234, 235, 238) ICP-MS and/or alpha spectrometry
- Po-210, Pb-210 alpha spectrometry, LSC





Results – Gamma spectrometry

27 clay samples											
		Activity Concentration [Bq/kg]									
	Ra-226	±	Th-232	±	K-40	±	- I-Index				
AVG	37	7	40	9	803	37	0.59				
Min	16	3	31	7	534	16	0.40				
Max	105	17	49	11	1127	105	0.81				

20 Manganese clay samples								
	Activity Concentration [Bq/kg]						lindov	
	Ra-226	±	Th-232	±	K-40	±	I-muex	
AVG	41	4	40	4	585	20	0.53	
Min	23	3	25	2	369	15	0.37	
Max	63	6	53	5	757	16	0.65	

Low I-indexes \rightarrow can be mixed according to choices

Z. Sas, J. Somlai, J. Jonas, G. Szeiler, T. Kovacs, Cs. Gyongyosi, T. Sydo: Radiological Survey of Hungarian Clays; Radon Emanation and Exhalation Influential Effect of Sample and Internal Structure Conditions, ROMANIAN JOURNAL OF PHYSICS 58:Number supplement (2013)

Z. Sas, J. Somlai, G. Szeiler, T. Kovacs: Radon emanation and exhalation characteristic of heat-treated clay samples, RADIATION PROTECTION DOSIMETRY 152:(1-3) pp. 51-54. (2012)



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Results - Optimal measurement conditions



Free exhalation state EXHALED (DECAY WITHIN MATERIAL) • If thickness << Diffusion length • emanated radon \approx exhaled 325 Free exhalation depends on: 300 Specific exhalation [mBq kg¹ h⁻¹] • Ra-226 content 275 250 Emanation coefficient 225 Amount of the sample 200 175 150 125 100 2 3 0 1 4 5 Sample diamater [cm]

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Results – Effects of heat-treatment

Radon exhaling capacity

- The radon exhalation depends on applied heat
- In case of 750 °C only 5 % of the initial
- Significant porosity changes
- Radon emanation is in direct proportion to radon exhalation under free exhalation state

Effective way to reduce exhalation capacity



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Results – Internal structure features

Porosity changes

- Remarkable porosity changes
- Mezopores closed
- Between 20 to 60nm pore diameter
- Great effect on the emanation
- On exhalation capacity as well



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Results – Leaching test



- CEN/TS 15364:2006 Characterization of waste
- MSZ-21470-50: Hungarian regulation for toxic elements, heavy metals and chrome(IV) in soil
- Tessier-extraction: 5 step sequential extraction

Measurements in progress...



Results – Leaching test

Chemical components

NORM nuclides

Element	700 °C	850 °C		Flomont	700 °C	850 °C			
Liement	mg/ k	mg/ kg (dry)		Liement	mg/ kg (dry)				
Са	704±23	680±11		Cd	<0.12	<0.12			
						0.33+0.02	Flamant	700 °C	850 °C
Mg	34±4	15±3	Cu 0.34±0.02		0.0010101	LIEITIEIT	mBq/g (dry)		
Na	183±13	151±9		Zn	1.51±0. 2	2.11±0. 2	Ra-226	<1.5	<1.5
Cd	215±21	118±11		Sb	<0.01	<0.01	Th 232	<5	< 5
total Cr	<0.30	<0.30		Ва	0.63±0.02	2.05±0. 2	Po/Pb 210	11±	11±
Pb	<0.25	<0.25		Se	<0.012	<0.012	U-238	7 ± 3	8±4
Hg	0.014±	<0.005		Мо	<0.05	<0.05			
Ni	<0.25	<0.25		Fe	22±2	21±2			
As	<0.009	0.016±0.06		Mn	18±1	15±1			
									EU-NORM2



Reuse of Manganese clay



Based on the results: any mixing ratio is accepted over 750 C burning temp

Outside application



Inside application







Measurements in progress...



Preparation of samples:

 burning 1050 C
 self-glazing, coating is not necessary

Leaching method:

- by strong alcohol (pálinka)
 ambient prameters (pH, C etc)
 exctraction time (over 1 month)
 - organoleptic tests (by myself)



Thank you for your attention! Děkuji vám za pozornost! Köszönöm a figyelmet!





